

**Prod.Name:** TerraCair Urea Solution, AUS 32  
**Manufacturer:** Terra Industries Inc.  
**Supplier:** GM of Canada - Oshawa  
**HMCS ID:** 1189736  
**SUC:** 16 - General Use

# MATERIAL SAFETY DATA SHEET

**Revision:** 17.May.2007  
**Effective:** 17.May.2007  
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## 1 PRODUCT AND COMPANY IDENTIFICATION

### PRODUCT INFORMATION

**Product Name:** TerraCair Urea Solution, AUS 32

**Recommended Use:**

SCR NOx Control

**External Keys:**

PMRV0957 PMRV  
9986284 Productive Materials  
CN36333 GM of Canada Approval Number  
88862659 Distributable Material (Part #)  
88863523 Distributable Material (Part #)  
Diesel Exhaust Primary Tradename - Distributable Material  
Fluid  
88863494 Distributable Material (Part #)  
88863495 Distributable Material (Part #)  
19286291 Distributable Material (Part #)  
10-4022 Distributable Material (Part #)  
19286292 Distributable Material (Part #)  
10-4023 Distributable Material (Part #)

### MANUFACTURER INFORMATION

**Manufacturer:** Terra Industries Inc.

**Address:**

Terra Centre - 600 Fourth Street USA Iowa 51101 Sioux City P.O. Box 6000 Mailing

**Communication Lines:**

Phone 712-277-1340 Information  
Internet www.terraindustries.com MSDS available  
Phone 800-424-9300 CHEMTREC (U.S.)  
Phone 613-996-6666 CANUTEC (Canada)

### SUPPLIER INFORMATION

**Supplier:** GM of Canada - Oshawa

**Address:**

1908 Colonel Sam Drive CAN ON L1H8P7 Oshawa Mailing

## 2 INGREDIENT INFORMATION

**Chemical Family:** Amide

### FORMULATION

**Chemical Characterization:**

Formula: CH<sub>4</sub>N<sub>2</sub>O + H<sub>2</sub>O

**Ingredients:**

<u>Chemical Name</u>	<u>CAS Number</u>	<u>Prefix</u>	<u>Value</u>	<u>Unit</u>	<u>Exposure Limits</u>
UREA	57-13-6	Range	31.8 - 33.2	% Wt	No
AMMONIA	7664-41-7	<	0.2	% Wt	Yes
Imidodicarbonic diamide	108-19-0	<	0.3	% Wt	No
WATER	7732-18-5	Range	67.7 - 66.3	% Wt	No

## 3 HAZARDS IDENTIFICATION

**Hazards Overview:**

### EMERGENCY OVERVIEW

Colorless liquid. With slight ammonia (pungent) odor. Reacts with sodium hypochlorite or calcium hypochlorite to form the explosive nitrogen trichloride. When heated, urea

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### 3 HAZARDS IDENTIFICATION

#### Hazards Overview:

releases ammonia and when heated to decomposition it emits toxic fumes of nitrogen oxides (NOx), ammonia, and cyanuric acid. Use water to control fires involving urea solution if water is compatible with burning material. Urea solution itself is non flammable.

#### Specific Hazards:

Primary Routes of Entry: Skin contact/absorption, eye contact, and vapor inhalation.

#### Specific Hazards (Routes Of Exposure):

<u>Exposure Routes</u>	<u>Exposure Duration</u>	<u>Observation</u>
Eye Contact	Acute	May cause irritation to eyes.
Skin Contact	Acute	May cause irritation to skin.

#### Medical Conditions Aggravated By Exposure:

No test data available.

#### Additional Health Hazard Data:

Ammonia and carbon dioxide vapors may accumulate in a confined space.

General Chronic Exposure: No test data available.

### 4 FIRST AID MEASURES

#### First Aid By::

Inhalation	If irritation develops move patient to fresh air and monitor. If cough or difficulty in breathing develops, evaluate for respiratory tract irritation. If trained to do so, administer supplemental oxygen if needed. If irritation, coughing, or difficulty in breathing persists the patient should be seen in a health care facility.
Skin Contact	If irritation occurs, flush exposed area with copious amounts of tepid water for at least 15 minutes followed by washing area thoroughly with soap and water. The patient should be seen in a health care facility if irritation or pain persists.
Eye Contact	Flush eyes with copious amounts of tepid water for at least 15 minutes. If irritation, pain, swelling, excessive tearing, or light sensitivity persists, the patient should be seen in a health care facility.
Ingestion	If conscious, give the patient large quantities of water to drink and induce vomiting. Seek medical attention.

### 5 FIRE FIGHTING MEASURES

#### Product Flammability:

Urea solution is not flammable.

#### Extinguishing Media:

Use water to extinguish a fire involving urea solution if water is compatible with the burning material.

#### Fire and Explosion Hazards:

At elevated temperature, urea solution may decompose to form cyanuric acid, ammonia, biuret, and/or nitrogen oxides.

#### Special Fire Fighting Procedures:

Positive pressure self-contained breathing apparatus (SCBA) should be used when there is a potential for inhalation of vapors and/or fumes.

Wear full fire fighting protective equipment that is appropriate for conditions.

Runoff from fire control or dilution water may cause pollution.

### 6 ACCIDENTAL RELEASE MEASURES

#### PRECAUTIONS IN CASE OF ACCIDENTAL RELEASE

##### Personal Precautions:

Keep unnecessary people away and isolate hazard area. Urea solution may be toxic to cattle (ruminants) when ingested if amount ingested is not controlled properly.

Generally, a small spill is one that involves a single, small package (i.e. up to a 55 gallon drum), small cylinder, or a small (non-continuing) leak from a large container. Spilled urea solution may cause slippery conditions.

##### Environmental Precautions:

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### 6 ACCIDENTAL RELEASE MEASURES

#### PRECAUTIONS IN CASE OF ACCIDENTAL RELEASE

##### Environmental Precautions:

Runoff may cause pollution.

#### SPILL OR LEAK PROCEDURES

##### Recovery:

Recover and use as fertilizer.

If disposal of product or contaminated by-products is necessary, follow guidelines set forth by local, state, and federal environmental agencies.

### 7 HANDLING AND STORAGE

#### HANDLING

##### Safe Handling Procedures:

Use proper personal protective equipment when working with or around urea solution. (See section 8).

#### STORAGE

##### Storage Conditions:

No unusual storage precautions are necessary.

### 8 EXPOSURE CONTROLS/PERSONAL PROTECTION

##### Engineering Measures:

If necessary to enter an area that contains urea solution, monitor for ammonia and oxygen content. Oxygen levels should be maintained between 19.5% and 23.5%, if outside of this range use appropriate precautions. Adequate ventilation should be supplied. Safety shower and eyewash fountain or at least 5 gallons of accessible clean water should be provided in a urea solution storage/handling area.

#### EXPOSURE LIMITS

##### Limit Values:

<u>Chemical Name</u>	<u>CAS Number</u>	<u>Type</u>	<u>Value</u>	<u>Specification</u>	<u>Source</u>
AMMONIA	7664-41-7	PEL-TWA	50ppm	-	OSHA - Permissible Exposure Limits (PELs)
AMMONIA	7664-41-7	GM OEG -TWA	25ppm	-	GM Occupational Exposure Guidelines (OEG)
AMMONIA	7664-41-7	GM OEG -STEL	35ppm	-	GM Occupational Exposure Guidelines (OEG)
AMMONIA	7664-41-7	TLV-TWA	25ppm	-	Threshold Limit Values (TLVs) - ACGIH
AMMONIA	7664-41-7	TLV-STEL	35ppm	-	Threshold Limit Values (TLVs) - ACGIH
AMMONIA	7664-41-7	State-STEL	35ppm	-	MICHIGAN
AMMONIA	7664-41-7	State-STEL	35ppm	-	NEW YORK
AMMONIA	7664-41-7	State-STEL	35ppm	-	TENNESSEE

#### PERSONAL PROTECTIVE EQUIPMENT

##### Personal Protective Equipment (PPE):

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### 8 EXPOSURE CONTROLS/PERSONAL PROTECTION

#### PERSONAL PROTECTIVE EQUIPMENT

##### Personal Protective Equipment (PPE):

**Respiratory Protection** Urea solution may pose an inhalation hazard in confined areas due to its ability to produce ammonia and carbon dioxide vapors. If ammonia vapors are present, protect as follows:  
<25 ppm: No protection required.  
25 to 35 ppm: Protection required if the daily TWA is exceeded.  
35 to 50 ppm: Protection required if exposed for more than 15 minutes  
50 to 250 ppm: Minimum of an air-purifying respirator equipped with ammonia canister(s) or cartridge(s).  
250 to 300 ppm: Minimum of a full-face air-purifying respirator equipped with ammonia canister(s) or cartridge(s).  
>300 ppm: A fresh air supply system must be used (i.e. positive pressure self contained breathing apparatus)

**Hand Protection** Chemically Impervious gloves should be worn.

**Eye Protection** It is recommended that safety glasses or goggles be used and if there is a potential for splashing liquid, a face shield should be used in conjunction with the safety glasses or goggles.

### 9 PHYSICAL AND CHEMICAL PROPERTIES

#### APPEARANCE

**Physical State:** Liquid.

**Color:** Colorless.

**Odor:** Slight ammonia odor (pungent)

#### PHYSICAL PROPERTIES

##### pH Value:

No test results

##### Vapor Pressure:

No test results

##### Vapor Density:

No test results

##### Density:

Density = 9.1 lb/gal

##### Specific Gravity:

Range 1.087 - 1.093 at 20 C.

##### Solubility:

= 100 %

##### Total Amount Of::

Percent Volatile by Volume No test results

##### Additional Chemical and Physical Data:

Refractive Index Range 1.3814 - 1.3843 at 20 C.

Molecular Weight: Not Applicable

Critical Temperature No test results

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### 9 PHYSICAL AND CHEMICAL PROPERTIES

#### PHYSICAL PROPERTIES

##### Additional Chemical and Physical Data:

Critical Pressure No test results

### 10 STABILITY AND REACTIVITY

#### STABILITY INFORMATION

**Stability Under Normal Conditions:** Stable

##### Incompatible Materials:

Reacts with sodium hypochlorite or calcium hypochlorite to form nitrogen trichloride that may explode spontaneously in air. Incompatible with sodium nitrite, phosphorus pentachloride, and nitrosyl or gallium perchlorate. Urea will form urea nitrate when mixed with nitric acid at low pH. Urea nitrate may become unstable and/or explosive under certain conditions.

##### Hazardous Polymerization:

Will not occur.

#### HAZARDOUS DECOMPOSITION

##### Reactions:

<u>Type of Reaction</u>	<u>Reaction Products</u>
Thermal Decomposition	At elevated temperature, urea solution may decompose to form cyanuric acid, ammonia, biuret, and/or nitrogen oxides.
Decomposition	Urea solution forms ammonia, cyanuric acid, biuret, and/or nitrogen oxides (NOx) upon decomposition.

### 11 TOXICOLOGICAL INFORMATION

#### SCIENTIFIC OBSERVATIONS

##### LETHAL LIMIT VALUES

##### Product Data:

<u>Exposure Routes</u>	<u>Type</u>	<u>Prefix</u>	<u>Value</u>	<u>Unit</u>	<u>Species</u>	<u>Comment</u>
Ingestion	LD50	Range	14300 - 15000	mg/kg	Rat	
Ingestion	LD50	Range	11500 - 13000	mg/kg	Mouse	
Ingestion	LD50	=	510	mg/kg	Cattle	
Skin Contact		=	40	'%	Rat	Repeated Dose Toxicity: NOAEL = 40% in ointment (24 wks; dermal)

#### CLASSIFICATION OF INGREDIENTS

##### Carcinogenicity:

NTP: Not Listed  
IARC: Not Listed  
OSHA: Not Regulated

### 12 ECOLOGICAL INFORMATION

#### ENVIRONMENTAL IMPACT

##### Comment:

Notify local health and wildlife officials and operators of any nearby water intakes of contamination or discharge into or leading to waterways.

#### ECOTOXICITY

##### Comment:

Acute Toxicity to Fish  
LC50 Barillius barna 9,100 mg/L (96 hr)

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### 12 ECOLOGICAL INFORMATION

#### ECOTOXICITY

**Comment:**

Acute Toxicity to Aquatic Invertebrates  
EC50 Daphnia magna >10,000 mg/L (DIN 38412 Part II; 24 hr)  
Toxicity to Aquatic Plants  
TT Scenedesmus quadricauda >10,000 mg/L (192 hr cell multiplication inhibition test)

Note: Data is for Urea  
Source: TFI Product Testing Program April 2003

### 13 DISPOSAL CONSIDERATIONS

**Waste Disposal Information:**

Urea solution is not listed by the Federal EPA as a hazardous waste. Consult state/provincial and local environmental agencies for acceptable disposal methods. Recover product for use as a fertilizer if possible.

### 14 TRANSPORT INFORMATION

**Comment:**

Urea solution is not listed by any U.S. or Canadian transportation authority as a hazardous material and as such, no specific information is available.

### 15 REGULATORY INFORMATION

#### LABELLING

**Hazard Codes:**

NFPA Health	1
NFPA Flammability	0
NFPA Reactivity	0

**Comment:**

NFPA assigned by General Motors Technical Review.

#### NATIONAL REGULATIONS

**SARA 311/312:** No

**SARA 313:** Yes

**Immediate Health:** No

**Delayed Health:** No

**Fire:** No

**Sudden Pressure Release:** No

**Reactive:** No

**Other Regulation:**

SARA 313:	SARA 313 inherited from ingredient list.
CERCLA Hazardous Substances List:	Not Listed
TSCA Inventory:	Listed

### 16 OTHER INFORMATION

**Comments:**

Additional Exposure Limits: GM Occupational Exposure Guidelines (OEG) and State-TWA's were provided by General Motors.